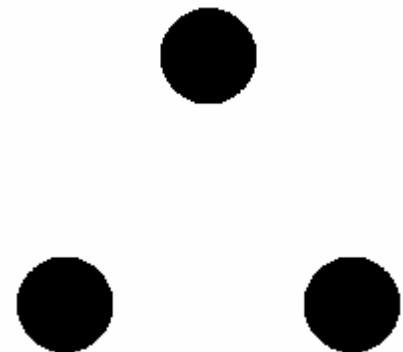




Laboratory spectroscopy of H₃⁺

Ben McCall

Oka Ion Factory™
University of Chicago



Motivations



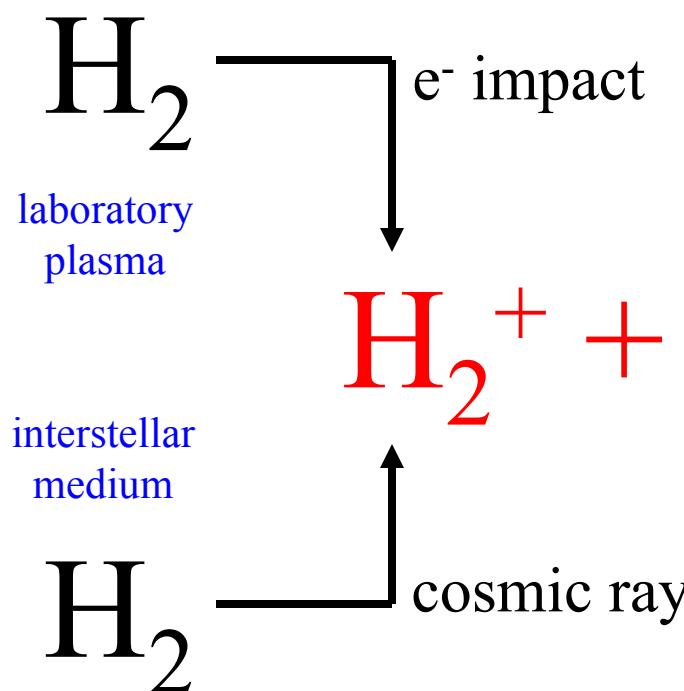
■ Astronomical

- obtain frequencies for detection & use as probe

■ Quantum Mechanical

- study structure of this fundamental ion
- refine theoretical calculations of polyatomics

Formation of H₃⁺



H₂ Proton Affinity:

~ 4.4 eV

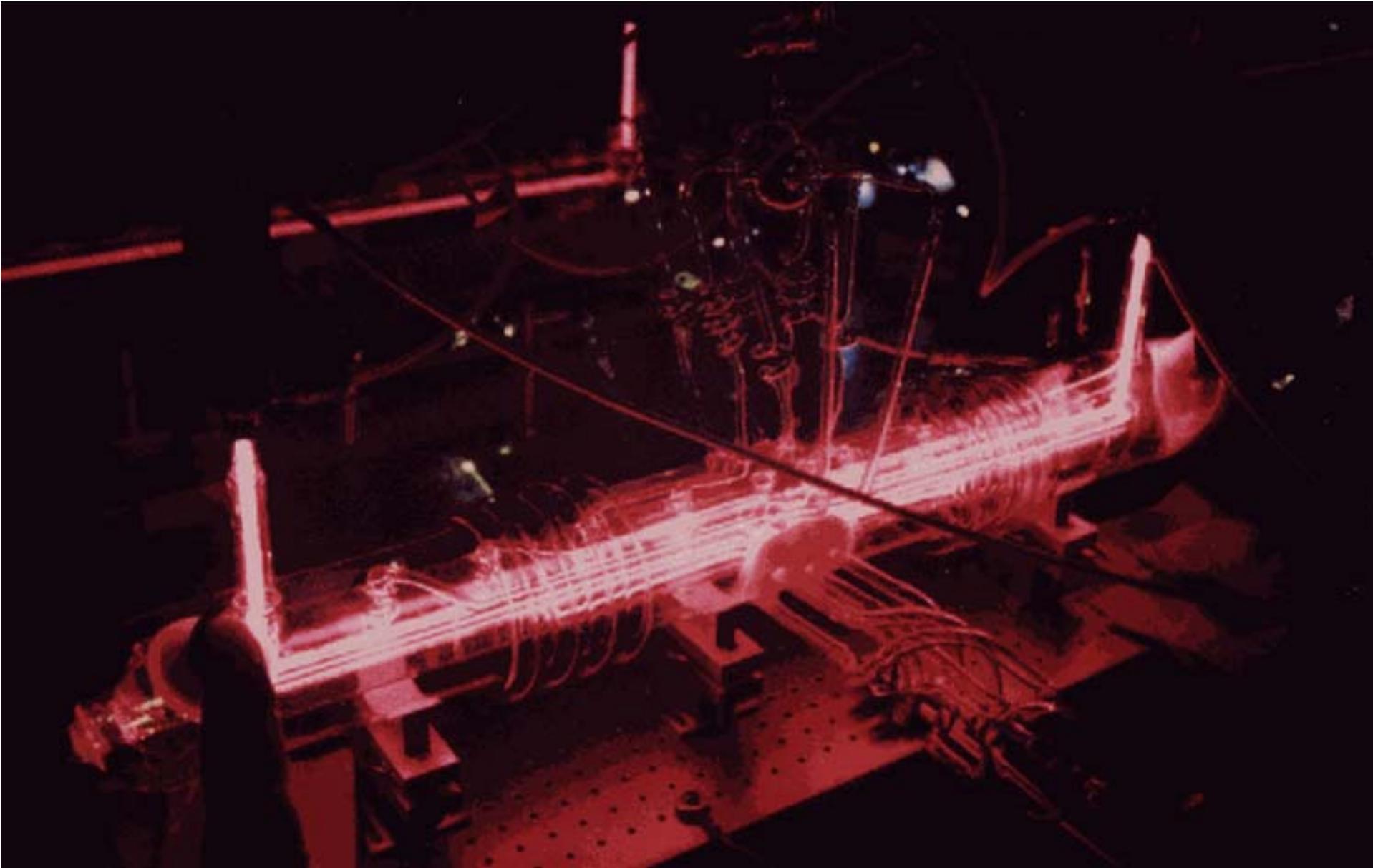
~ 100 kcal/mol

~ D(H₂)

Rate constant: k ~ 2 × 10⁻⁹ cm³ s⁻¹

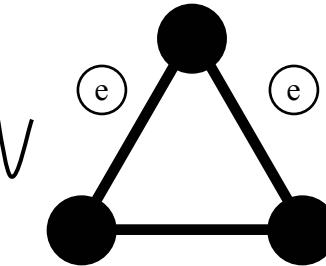
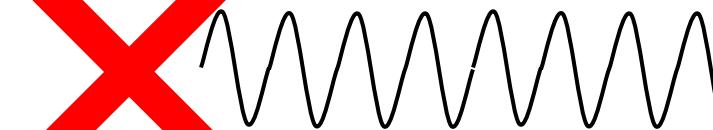
Exothermicity: ΔH ~ 1.7 eV

Laboratory Plasma



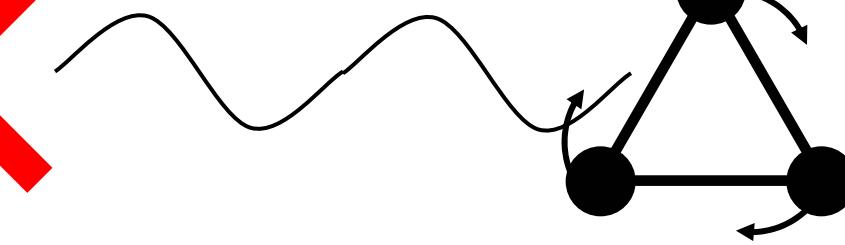
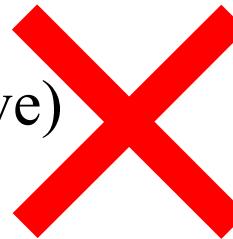
Types of Molecular Spectroscopy

Electronic (visible)

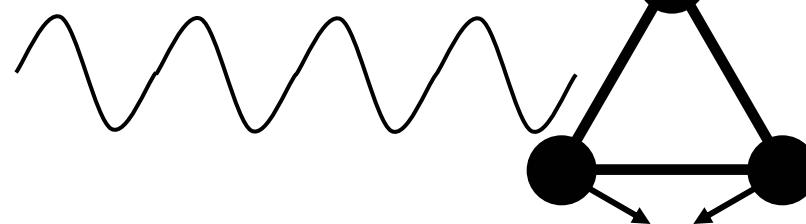


Poster: Friedrich & Alijah

Rotational (microwave)

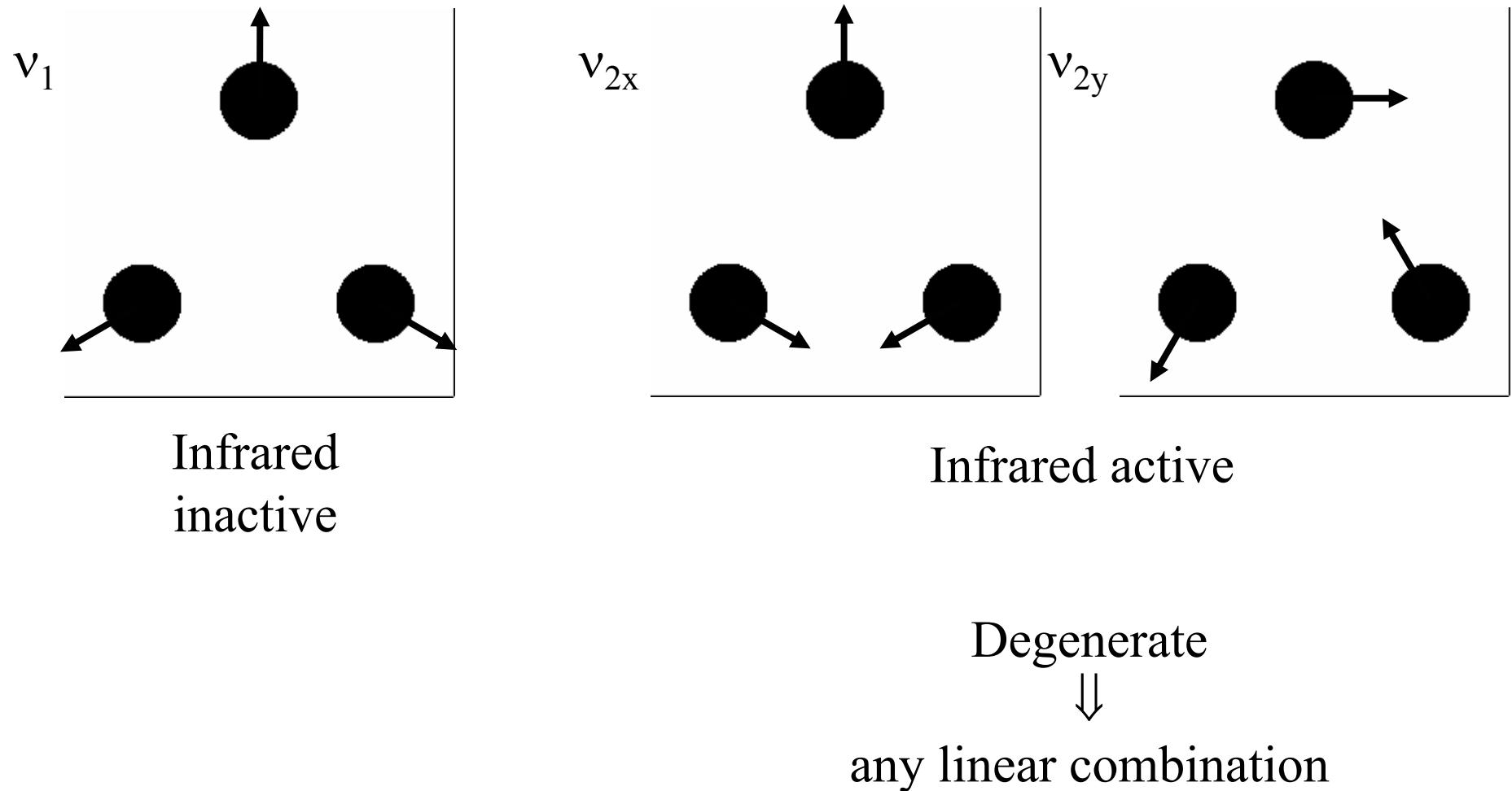


Vibrational (infrared)

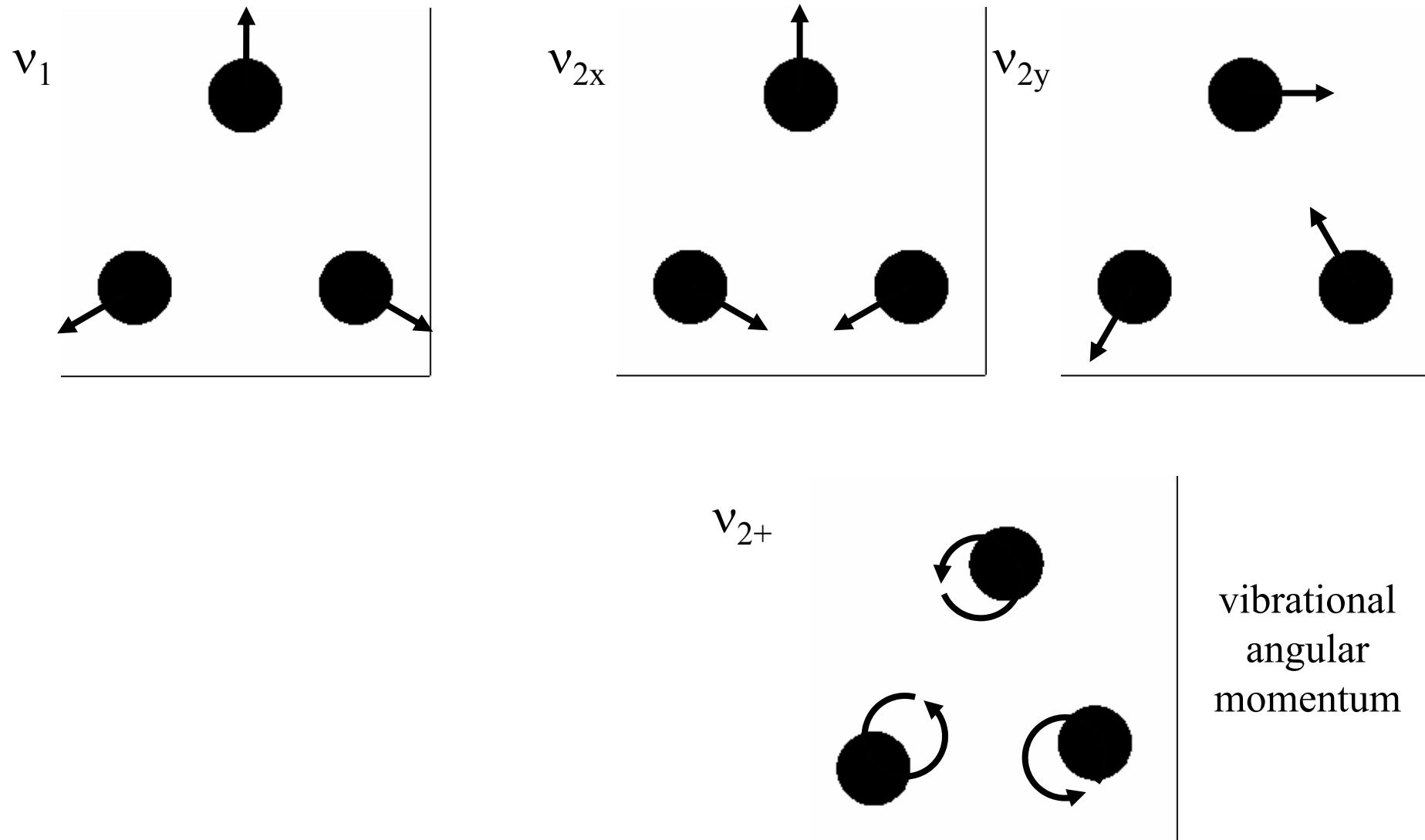


Ben McCall

Vibrational Modes of H_3^+



Vibrational Modes of H_3^+

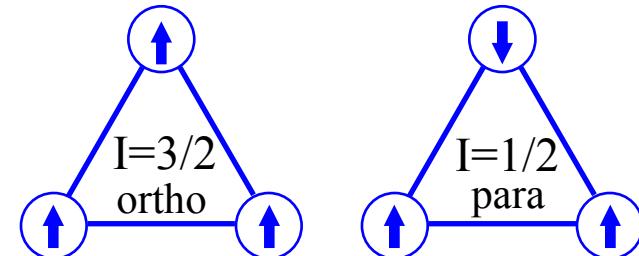


Quantum Numbers for H_3^+



■ Angular momentum

- I — nuclear spin
- J — nuclear motion
 - » $k = \text{projection of } J$; $K = |k|$
- \mathbf{l}_2 — vibration



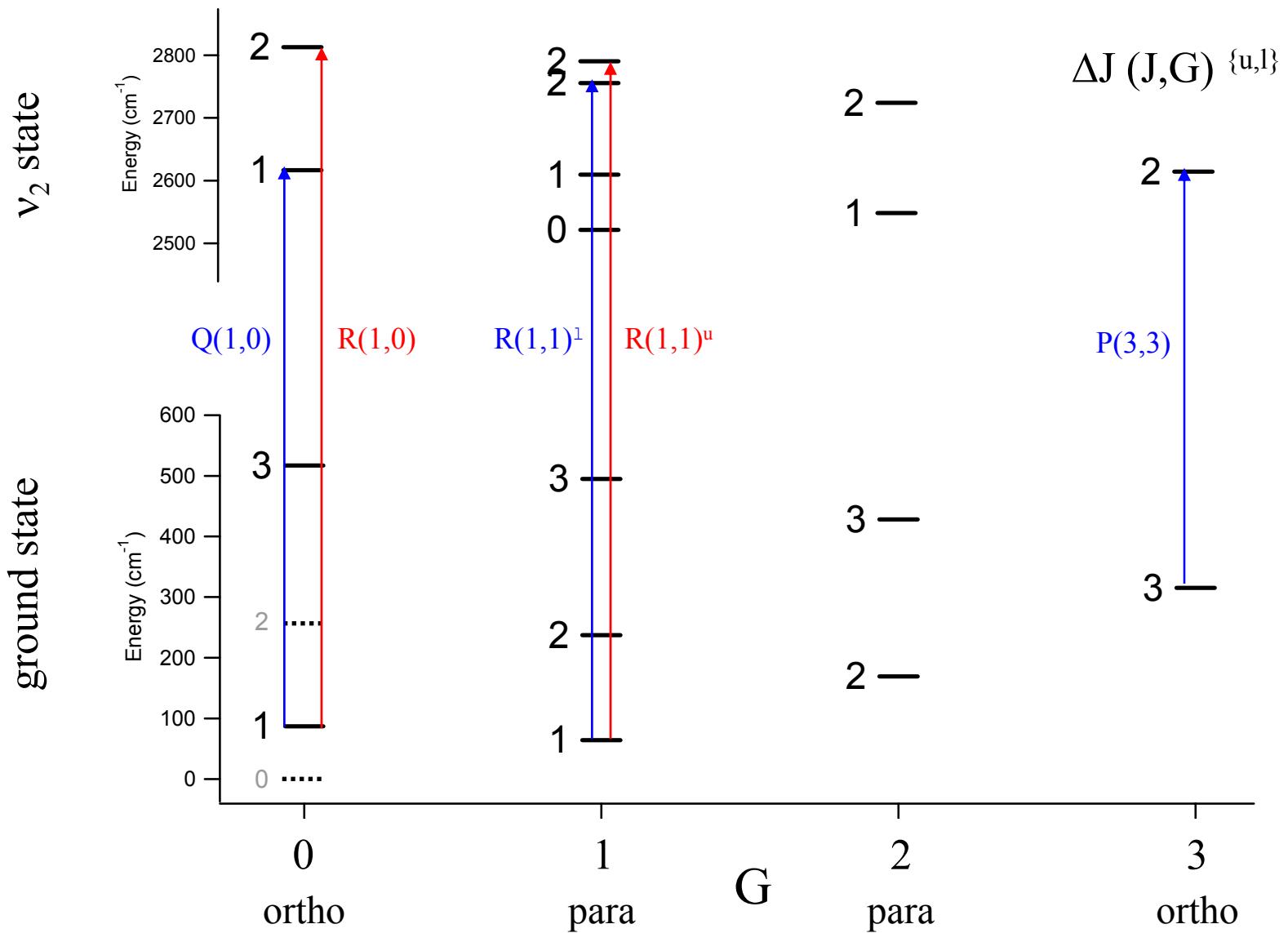
■ “ \mathbf{l} -resonance”: states with same $|\mathbf{k}-\mathbf{l}_2|$ mixed

- $G \equiv |\mathbf{k}-\mathbf{l}_2|$ — rotational part of J

■ Symmetry requirements

- $G=3n \leftrightarrow \text{ortho}$, $G \neq 3n \leftrightarrow \text{para}$
- Pauli: certain levels forbidden ($J=\text{even}$, $G=0$)

The $v_2 \leftarrow 0$ fundamental band

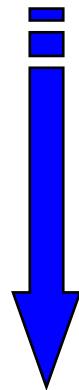


Experimental work on $\nu_2 \leftarrow 0$



■ Initial Detection (Oka 1980)

- search over two years, over 500 cm^{-1}
- 15 lines, few percent deep
- assigned by Watson overnight
- $\nu_2 = 2521\text{ cm}^{-1}$



Oka
1980

Oka
1981

Watson
et al. 1984

Majewski
et al. 1987

Nakanaga
et al. 1990

Uy
et al. 1994

Majewski
et al. 1994

McKellar &
Watson 1998

Joo
et al. 2000

Lindsay
et al. 2000
Ben McCall

15

30

46

J=9
FTIR
emission
113

FTIR
absorption
151

J=15
151

206
FTIR
emission

FTIR absorption
wide-band

207
Lowest frequency
 1546.901 cm^{-1}

217
Continuous scan
 $3000 - 3600\text{ cm}^{-1}$

The $v_2 \leftarrow 0$ band as a probe of H_3^+



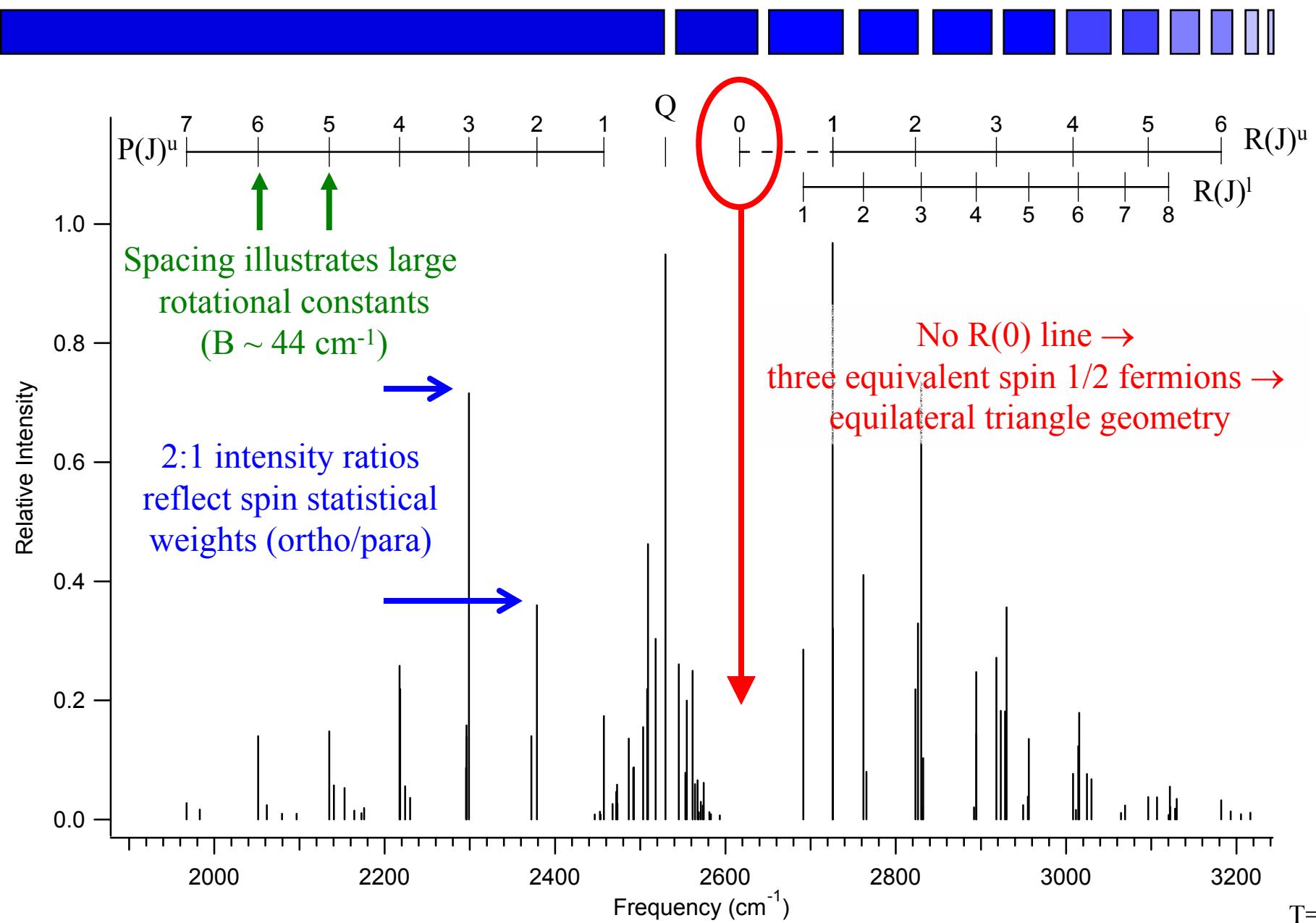
■ Laboratory

- H_3^+ electron recombination rate (Amano 1988)
- spin conversion of H_3^+ in reactions (Uy et al. 1997)
- ambipolar diffusion in plasmas (Lindsay, poster)

■ Astronomy

- probe of planetary ionospheres (Connerney, Miller)
- confirmation of interstellar chemistry (Herbst)
- measurements of interstellar clouds (Geballe)

The fundamental band

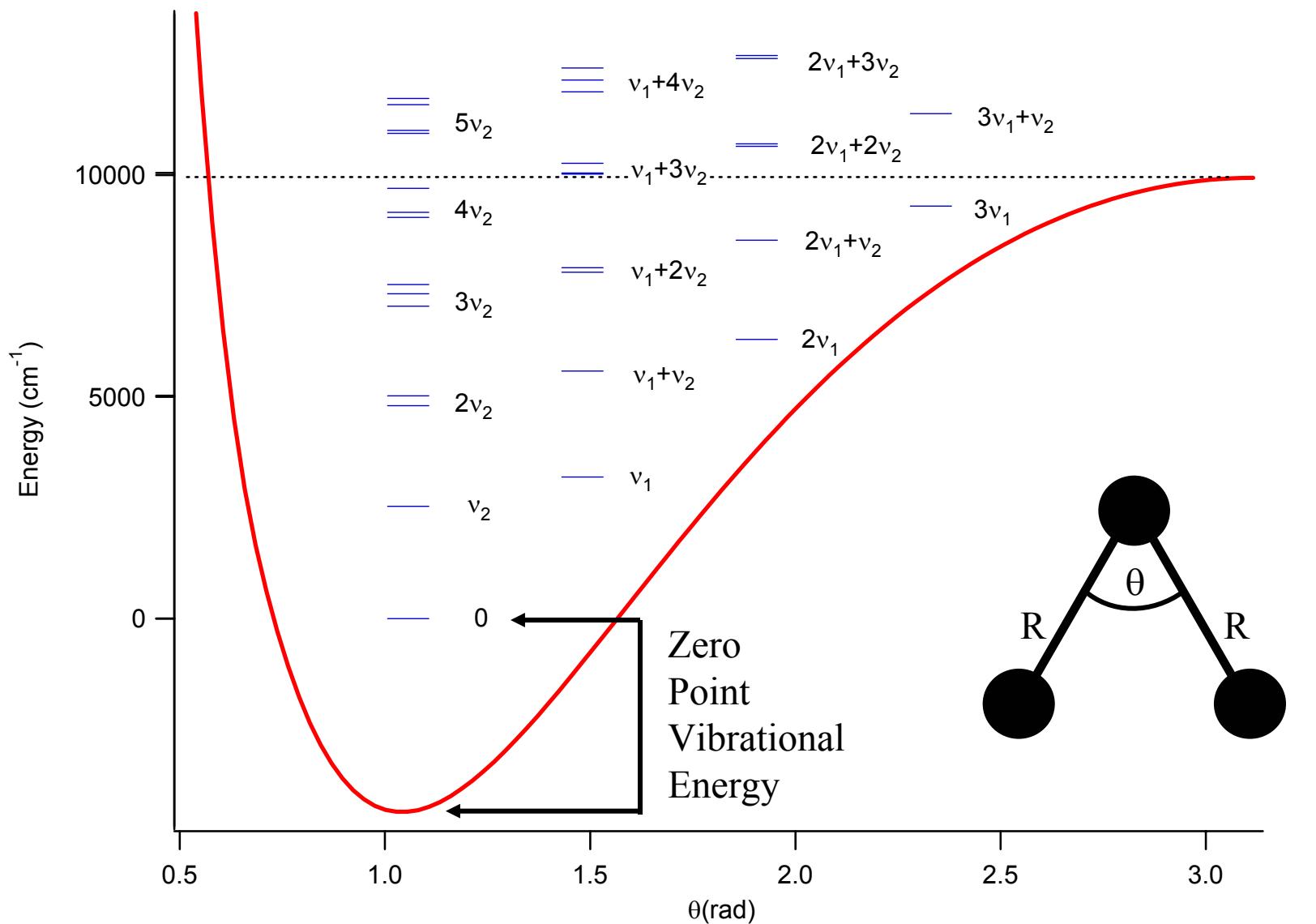


Understanding the ν_2 Spectrum

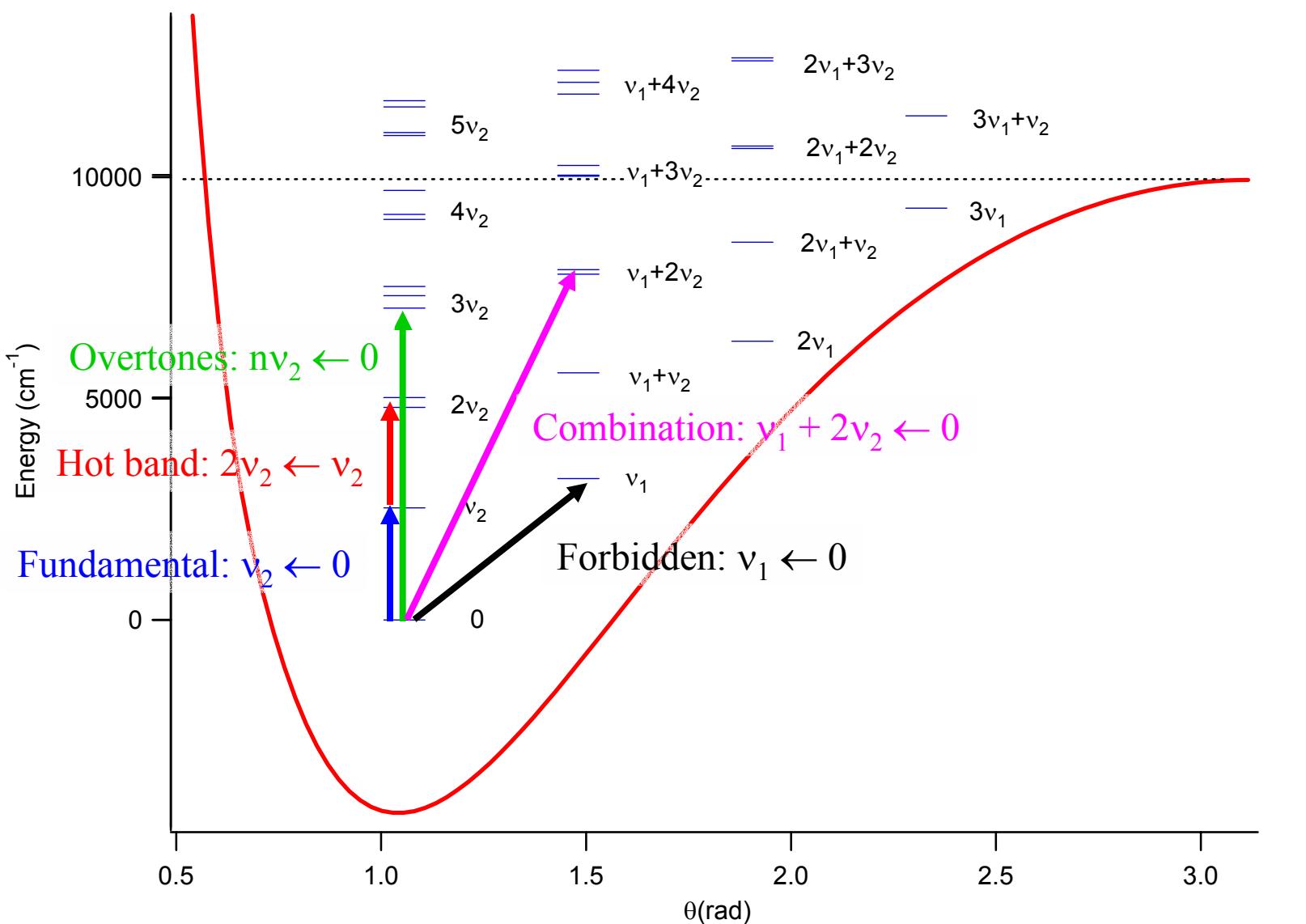


- For low energies, use perturbation approach:
 - $E'' = BJ(J+1) + (C-B)K^2 - D_{JK}J(J+1)K^2 + \dots$
 - $E' = \nu_2 + B'J'(J'+1) + (C'-B')K'^2 - 2\zeta C'K'\perp + \dots$
 - use observed transitions to fit molecular constants
 - For higher vibrational energies, this approach completely breaks down
- Variational calculations based on an *ab initio* potential energy surface!

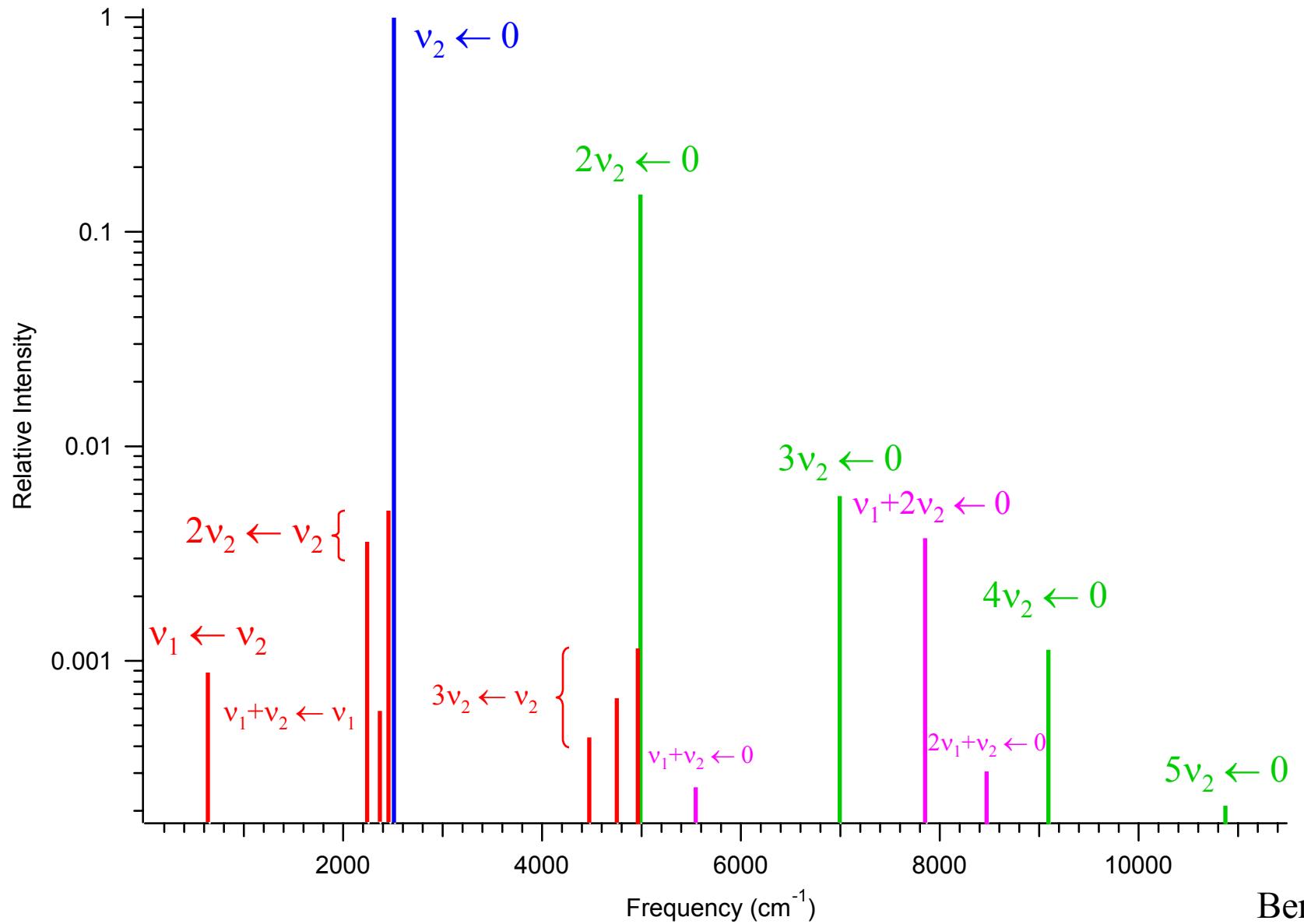
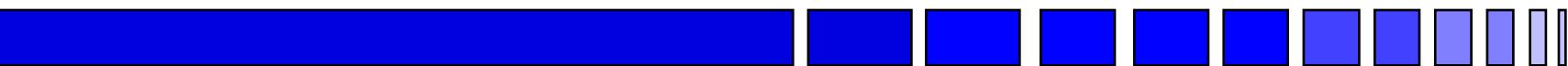
Variational Calculations



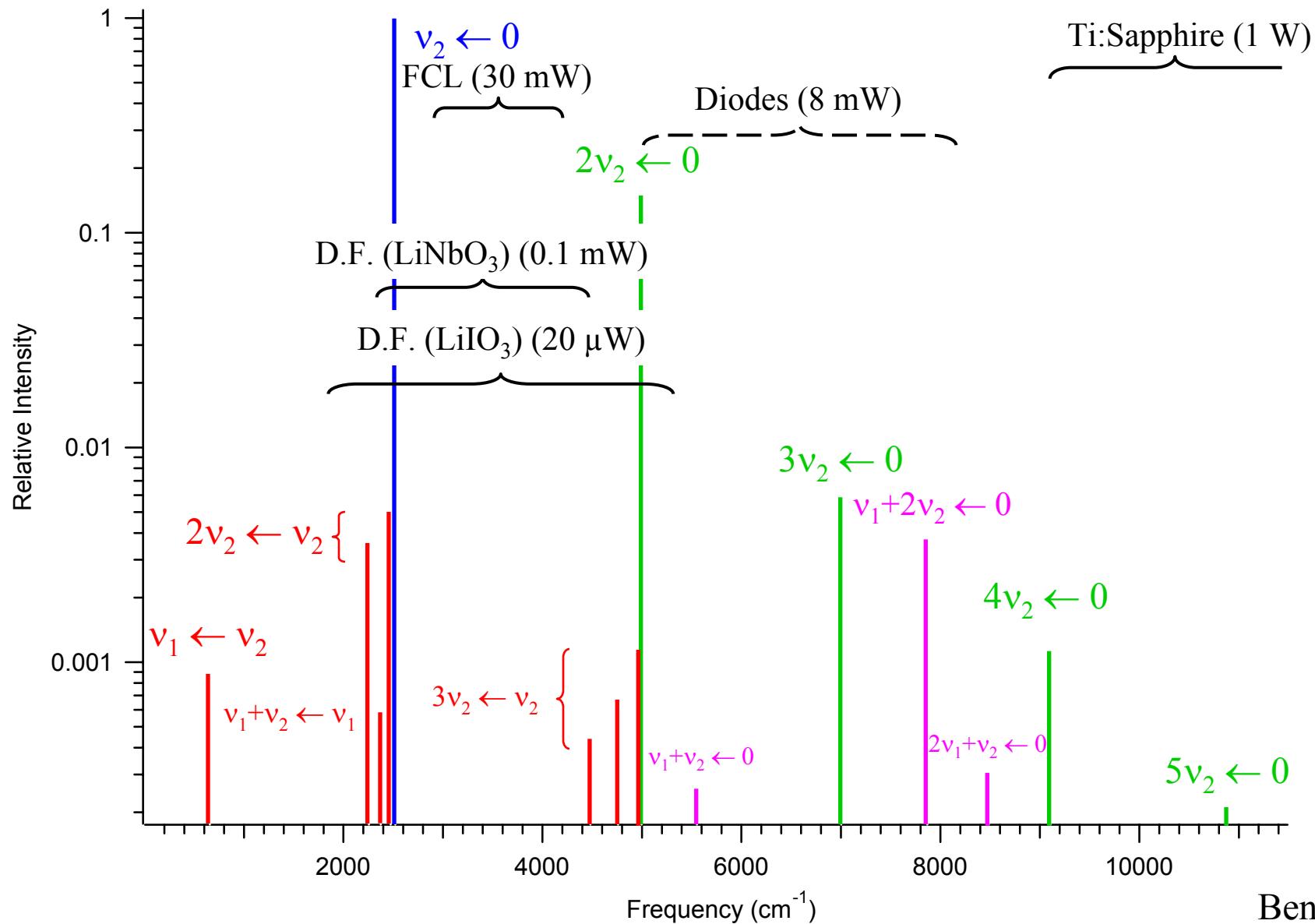
Vibrational Band Types



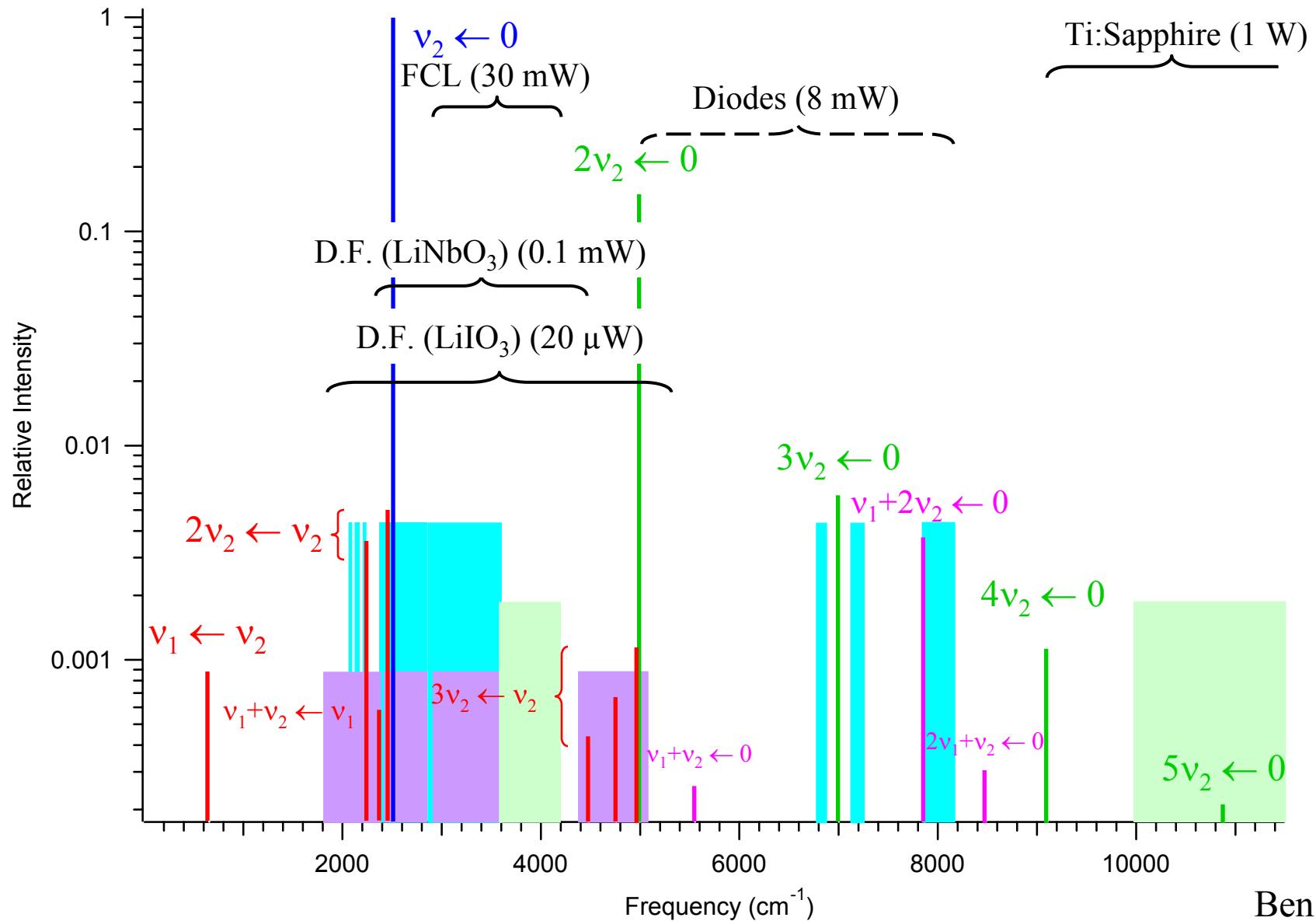
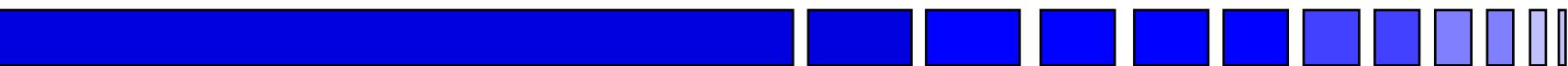
Vibrational Overview



Vibrational Overview



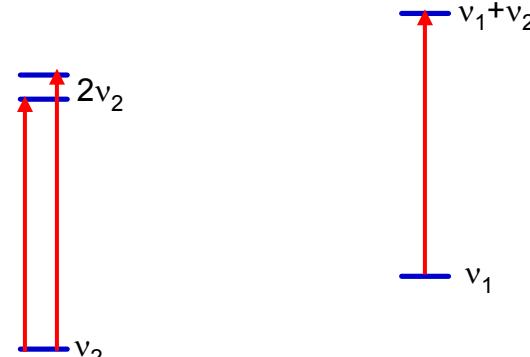
Vibrational Overview



Hot Bands



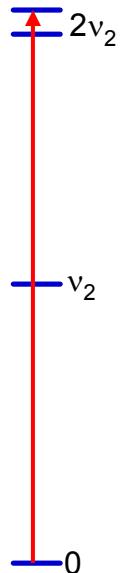
- At 600 K, \sim 200 times weaker than fundamental
- He dominated discharge \rightarrow only 50 times weaker
- Bawendi et al. (1990)
 - 72 lines of $2\nu_2^2 \leftarrow \nu_2$
 - 14 lines of $2\nu_2^0 \leftarrow \nu_2$
 - 21 lines of $\nu_1 + \nu_2 \leftarrow \nu_1$
- Variational calculations essential in assignment
 - Sutcliffe 1983; Miller & Tennyson 1988, 1989



First Overtone Band: $2v_2 \leftarrow 0$



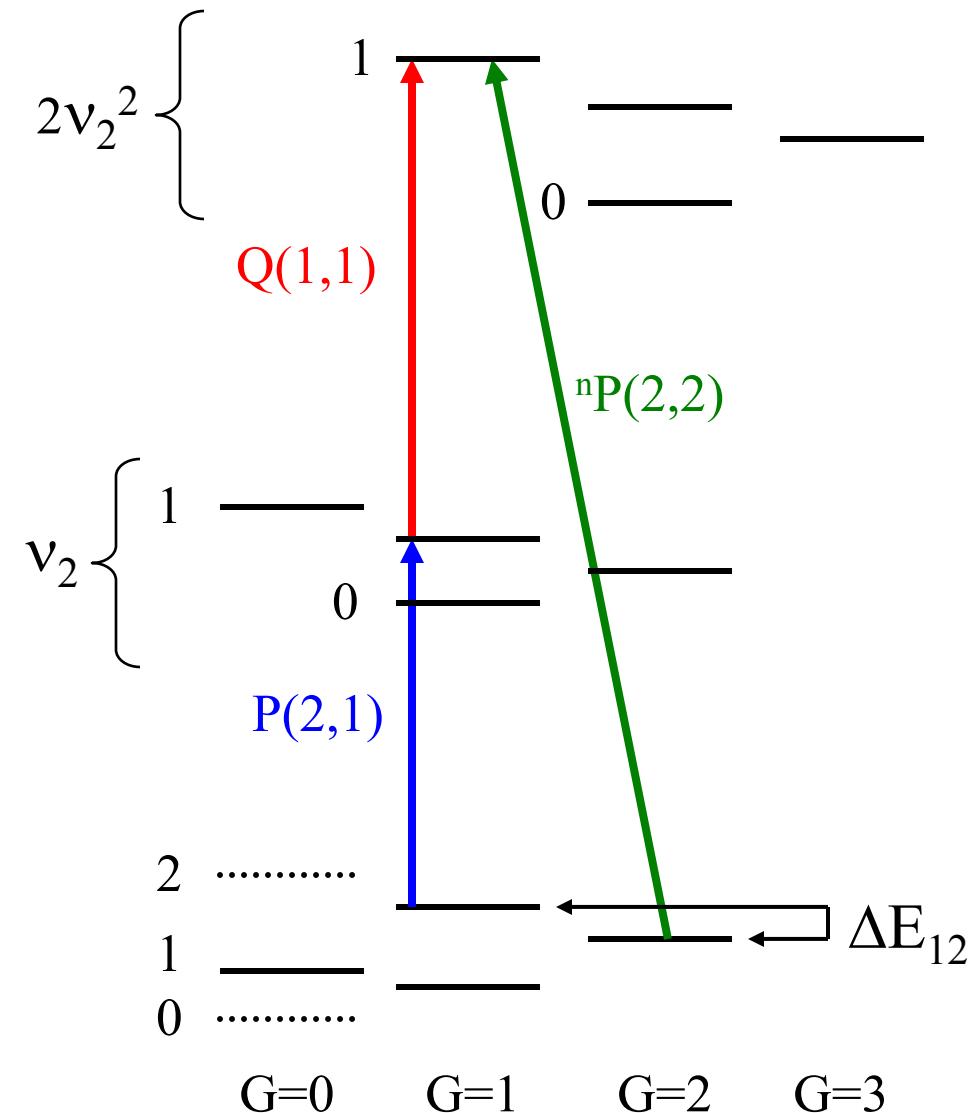
- First overtone ($2v_2 \leftarrow 0$) usually orders of magnitude weaker than fundamental
- In H_3^+ , only about 7 times weaker
- Discovery:
 - (in hindsight) Majewski et al. (1987)
 - Jupiter (Trafton et al. 1989, Drossart et al. 1989)
 - Assigned by Watson (with aid of hot bands)
 - Majewski et al. (1989) – 47 transitions, FTIR
 - Xu et al. (1990) – transitions observed in absorption



Absolute Energy Levels



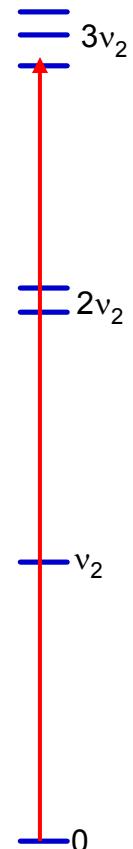
- Fundamental: $\Delta G = 0$
- Hot bands: $\Delta G = 0$
- Overtone band: $\Delta G = \pm 3$
 - ($n \leftrightarrow \Delta G = -3$, $t \leftrightarrow \Delta G = +3$)
- Absolute energy levels



Second Overtone Band: $3\nu_2 \leftarrow 0$



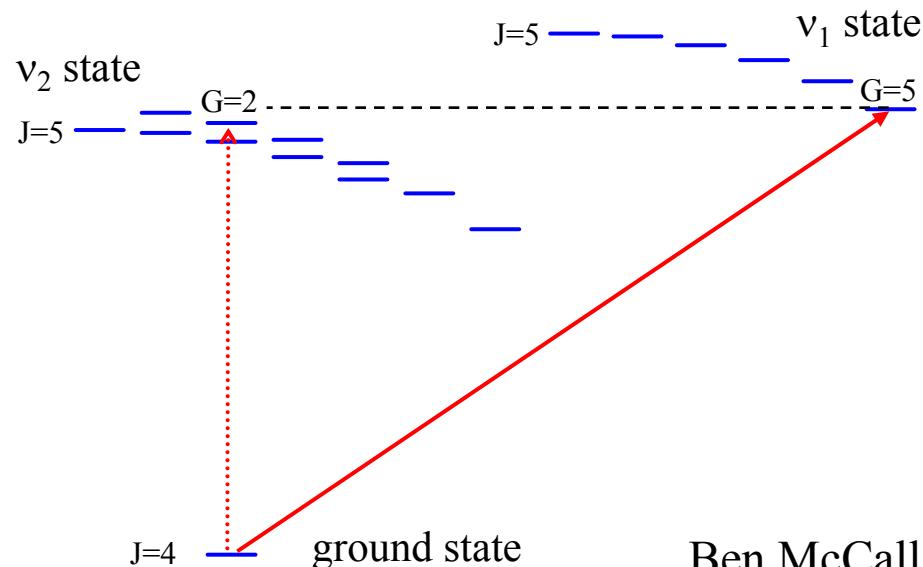
- ~ 200 times weaker than fundamental
- Band origin $\sim 7000 \text{ cm}^{-1}$
- Tunable diode lasers
- Lee et al. (1991), Ventrudo et al. (1994)
- 15 transitions observed
- Assigned based on variational calculations
 - Miller & Tennyson (1988, 1989)



Forbidden Band $\nu_1 \leftarrow 0$



- ν_1 mode totally symmetric \rightarrow infrared inactive
- $\nu_1 \leftarrow 0$ very forbidden since ν_1 & ν_2 not coupled
- First mixing term: “Birss resonance”
 - mixes levels in ν_1 & ν_2 with same J , $\Delta G=3$
 - effective for accidental degeneracies (fairly high J)
- Xu et al. (1992)
 - 9 lines
 - $\nu_1 = 3178 \text{ cm}^{-1}$
- Lindsay et al. (2000)
 - 10 new lines

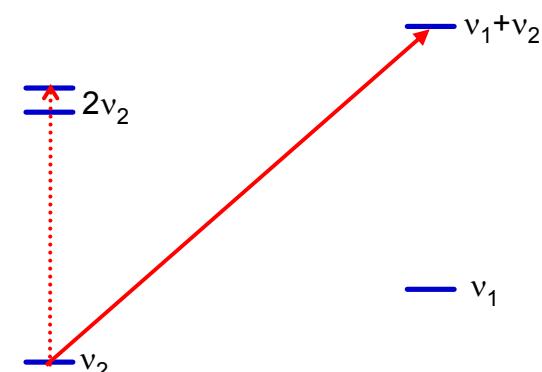


Forbidden Band $v_1 + v_2 \leftarrow v_2$



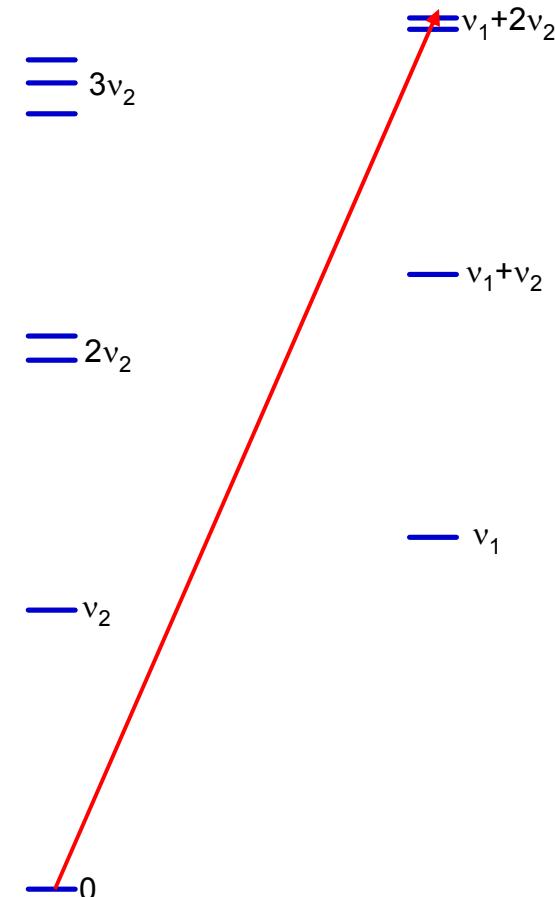
- Not as forbidden as $v_1 \leftarrow 0$
 - anharmonicity of potential mixes $v_1 + v_2$ with other states (e.g. $2v_2^2$)
 - $v_1 + v_2 \leftarrow v_2$ can “borrow” intensity from allowed bands (e.g. $2v_2^2 \leftarrow v_2$)

- Xu et al. (1992) observed 21 lines



Combination Bands

- $v_1 + 2v_2^2 \leftarrow 0$
- Highest energy band
- Weakest allowed band
 - 270 times weaker than v_2
- Tunable diode laser
 - $7780 - 8168 \text{ cm}^{-1}$



Observed Spectrum

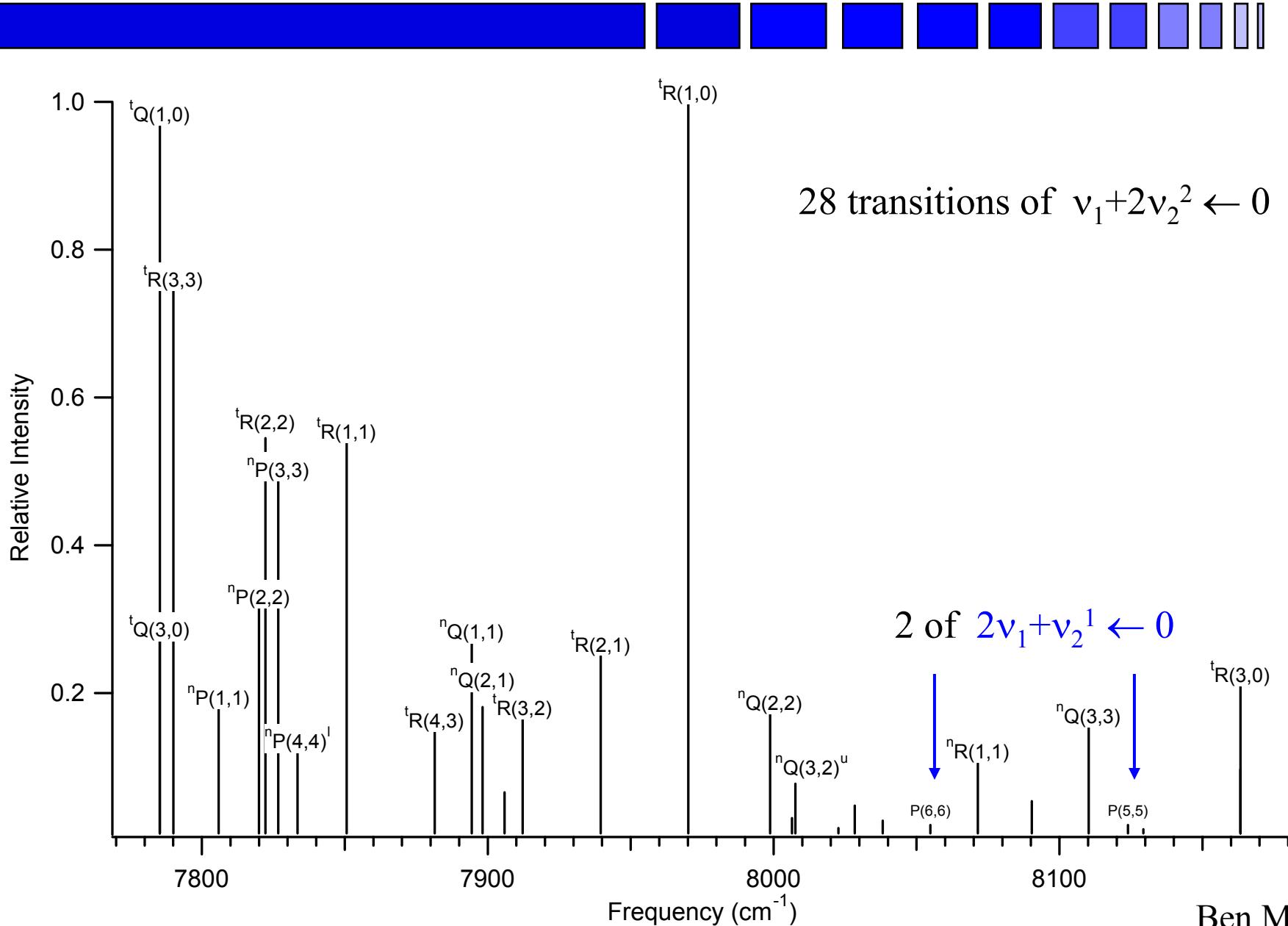


Table of Results

Symbol	Obs	Calc	O-C	J'	$\langle G' \rangle$	\pm	o/p	n'	$\langle v1' \rangle$	$\langle v2' \rangle$	$\langle l2' \rangle$	J"	k"
^t Q(3,0)	7785.233	7785.264	-0.032	3	3.0	-	o	12	1.0	2.0	-2.0	3	0
^t Q(1,0)	7785.701	7785.340	0.361	1	3.0	-	o	5	1.0	2.0	-2.0	1	0
^t R(3,3)	7789.878	7790.025	-0.147	4	5.9	+	o	9	1.0	2.0	-2.0	3	3
ⁿ P(1,1)	7805.893	7805.851	0.043	0	2.0	+	p	5	1.0	2.0	2.0	1	1
ⁿ P(2,2)	7820.239	7819.980	0.259	1	1.0	-	p	12	1.0	2.0	2.0	2	2
^t R(2,2)	7822.375	7822.250	0.125	3	5.0	-	p	18	1.0	2.0	-2.0	2	2
ⁿ P(3,3)	7826.739	7826.679	0.060	2	0.0	+	o	6	1.0	2.0	2.0	3	3
ⁿ P(4,4) ^l	7833.249	7833.444	-0.195	3	1.1	-	p	21	1.0	2.0	1.8	4	4
^t R(1,1)	7850.959	7850.677	0.283	2	4.0	+	p	15	1.0	2.0	-2.0	1	1
^t R(4,3)	7880.921	7881.443	-0.522	5	5.9	+	o	13	1.0	2.0	-2.0	4	3
ⁿ Q(1,1)	7894.711	7894.378	0.333	1	2.0	+	p	5	1.0	2.0	2.0	1	1
ⁿ Q(2,1)	7898.371	7898.187	0.183	2	2.0	+	p	17	1.0	2.0	2.0	2	1
ⁿ Q(3,1)	7905.717	7905.891	-0.174	3	2.0	+	p	17	1.0	2.0	1.9	3	1
^t R(3,2)	7912.047	7912.196	-0.148	4	4.9	-	p	18	1.0	2.0	-2.0	3	2
^t R(2,1)	7939.619	7939.499	0.120	3	4.0	+	p	15	1.0	2.0	-2.0	2	1
^t R(1,0)	7970.413	7970.124	0.288	2	3.0	-	o	5	1.0	2.0	-2.0	1	0
ⁿ Q(2,2)	7998.890	7998.754	0.136	2	1.0	-	p	12	1.0	2.0	2.0	2	2
^t R(4,2)	8005.582	8006.441	-0.858	5	4.8	-	p	30	1.0	2.0	-1.9	4	2
ⁿ Q(3,2) ^u	8007.410	8007.628	-0.218	3	1.0	-	p	22	1.0	2.0	1.9	3	2
ⁿ Q(4,2) ^u	8022.012	8022.693	-0.681	4	1.0	-	p	22	1.0	2.0	1.7	4	2
^t R(3,1)	8027.840	8028.337	-0.497	4	3.5	+	p	26	1.0	2.0	-1.8	3	1
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P(6,6)	8053.382	8054.810	-1.428	5	5.5	-	o	17	1.8	1.2	-1.2	6	6
ⁿ R(1,1)	8071.617	8071.414	0.203	2	2.0	+	p	17	1.0	2.0	2.0	1	1
ⁿ Q(4,3)	8089.406	8090.282	-0.876	4	0.1	+	o	11	1.0	2.0	2.0	4	3
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P(5,5)	8123.128	8123.985	-0.857	4	4.8	+	p	30	1.9	1.1	-1.1	5	5
^t R(4,1)	8128.280	8129.331	-1.051	5	3.4	+	p	27	0.9	2.1	-1.8	4	1
ⁿ R(2,1)	8163.129	8163.294	-0.165	3	2.0	+	p	17	1.0	2.0	1.9	2	1
^t R(3,0)	8162.653	8163.319	-0.666	4	2.9	-	o	12	1.0	2.0	-1.9	3	0

Predictions
from
J. K. G. Watson

Ben McCall

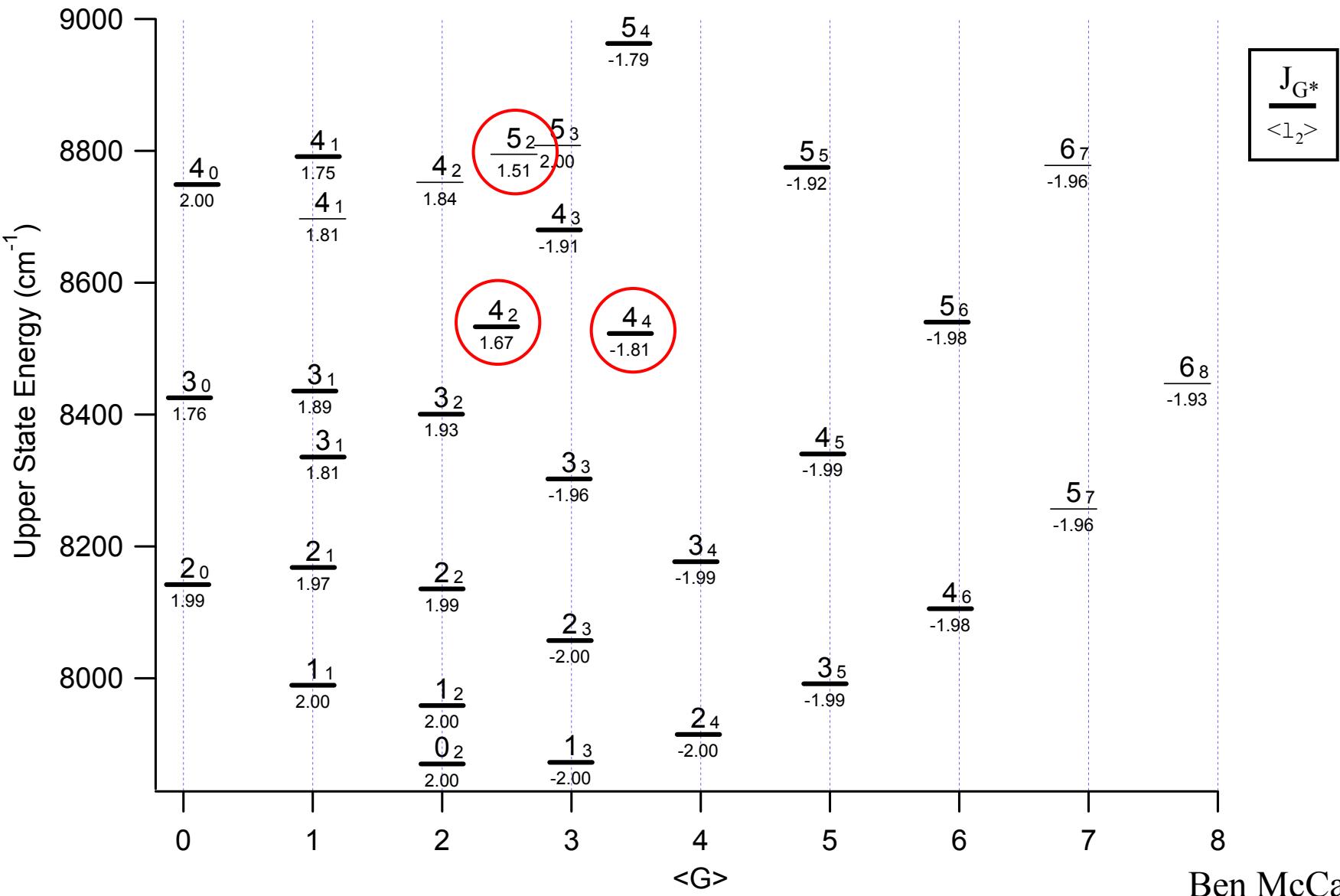
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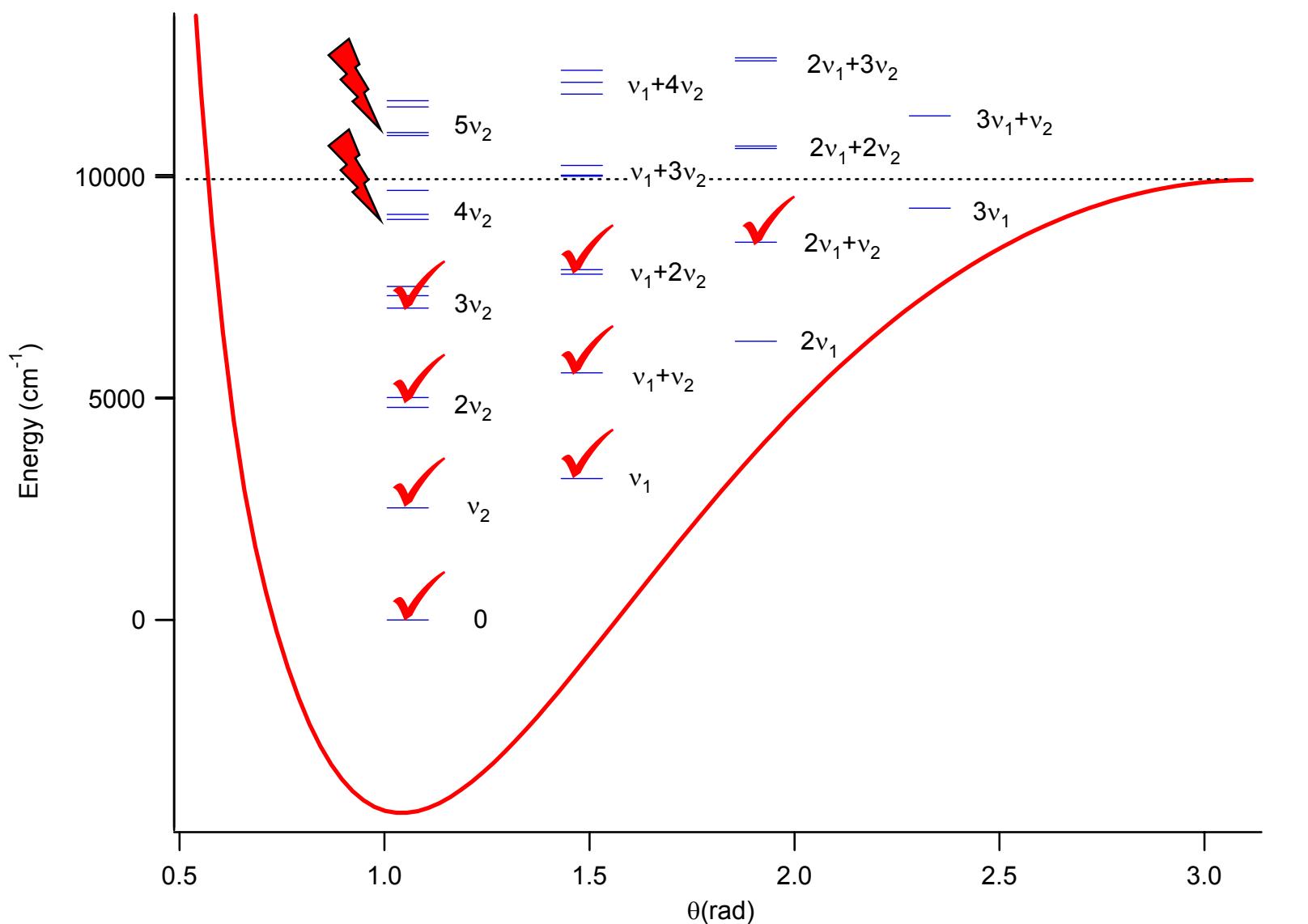
Predictions
from
J. K. G. Watson

Ben McCall

$\nu_1 + 2\nu_2$ Energy Levels



Breaking the Barrier to Linearity



Future Prospects



■ Improvements in Theory

- Hyperspherical coordinates for linearity (Watson)
- Relativistic effects (Jaquet)
- Non-adiabatic effects (Polyansky & Tennyson 1999)

■ Experimental Advances

- Titanium:Sapphire laser, Dye laser
- New techniques (heterodyne, cavities?)
- $4v_2 \leftarrow 0, 5v_2 \leftarrow 0$ on the horizon
- ... $10v_2 \leftarrow 0$ in the future??

Acknowledgements



- Oka
- J. K. G. Watson
- Fannie and John Hertz Foundation
- NSF
- NASA

Column Density of H₃⁺



(concentration) \times (path length) \equiv Column Density

Laboratory: 10^{11} cm^{-3} \times 10^3 cm $=$ 10^{14} cm^{-2}

Molecular
Cloud: 10^{-4} cm^{-3} \times 10^{18} cm $=$ 10^{14} cm^{-2}

Laboratory and astronomical spectroscopy progressing together!